

NANOCHEMISTRY

| Semester | Subject Code | Category | Lecture Hours | | Theory hours | | Practical hours | | Credits |
|----------|--------------|------------------------------|---------------|----------|--------------|----------|-----------------|----------|---------|
| | | | Per week | Per sem. | Per week | Per sem. | Per week | Per sem. | |
| VI | 21CCH6Ca | Elective - III (Option-2) | 4 | 60 | 4 | 60 | - | - | 3 |

COURSE OBJECTIVES:

The students will be able to

- Know the basics of nanotechnology, classification of nanomaterials and applications of nanotechnology in medicine, defence, agriculture and consumer products.

COURSE OUTCOMES:

On the successful completion of the course, students will be able to

| CO Number | CO Statement | Knowledge Level (K1-K4) |
|-----------|--|-------------------------|
| CO1 | Learn about the basics of Nanochemistry and classification of nanomaterials. | K2 |
| CO2 | Learn about the various properties of nanomaterials. | K4 |
| CO3 | Learn about different methods for the synthesis of nanoparticles. | K2 |
| CO4 | Learn about the nanowires, nanoclusters and their applications. | K3 |
| CO5 | Learn about nanotubes and the general applications of nanomaterials. | K3 |

*CO – Course Outcomes

Knowledge Level: K1 – Remember; K2 – Understand; K3 – Apply; K4 – Analyze.

MAPPING WITH PROGRAMME OUTCOMES:

| COS | PO1 | PO2 | PO3 | PO4 | PO5 | PO6 |
|-----|-----|-----|-----|-----|-----|-----|
| CO1 | S | M | M | S | M | S |
| CO2 | S | M | S | S | S | S |
| CO3 | S | S | S | S | S | S |
| CO4 | S | M | S | S | S | S |
| CO5 | S | S | S | S | S | S |

(S – Strong; M – Medium; L – Low)

UNIT-I: Basics of Nanochemistry **12 Hours**

- 1.1 Introduction – Basic Terminology– Particle, nano, nanometer, nanoparticle, nanomaterial, nanochemistry and nanotechnology (only definition).
- 1.2 Classification of Nanomaterials – Zero (0-D), One (1-D), Two (2-D) and Three (3-D) Dimensional (dimensions, criteria and examples) – Molecules, nanoparticles and bulk materials (differences).

UNIT-II: Properties of Nanomaterials **12 Hours**

- 2.1 Properties of Nanomaterials – Thermal, electrical, mechanical, optical, chemical and magnetic properties.
- 2.2 Size dependent properties – Property and influence of size reduction on properties of nanoparticles – Structural, mechanical, thermal, thermodynamical, electrical, magnetic, optical and chemical properties.

UNIT-III: Synthesis of Nanoparticles **12 Hours**

- 3.1 Synthesis of Nanoparticles – Top-Down Approach – Laser ablation, Chemical vapour Deposition – Hot Wire CVD, Plasma Enhanced CVD, Atomic Layer CVD and Metal Organic CVD and Electrode position.
- 3.2 Bottom-up Approach – Precipitation and thermolysis – Hydrothermal method and solvothermal method.

UNIT-IV: Nanowires and Nanoclusters **12 Hours**

- 4.1 Nanowires – Definition, characterizations, types of nanowires – Synthesis of nanowires – Solution – Phase Synthesis, template assisted synthesis and VSL Method (Vapour-Liquid –Solid)–Applications of nanowires – Uses of nanowires.
- 4.2 Nanoclusters – Definition – Characterisation – Magic number –Source of nanoclusters – Seeded supersonic nozzle sources and Gas aggregation cluster sources – Uses.

UNIT-V: Nanotubes and Applications of Nanomaterials **12 Hours**

- 5.1 Nanotubes – Carbon nanotubes - Types of nanotubes – Single walled carbon nanotubes – zig-zag nanotubes, chiral nanotubes and armchair nanotubes – Multi- walled carbon nanotubes .

5.2 Applications of nanomaterials in energy production, defence, agriculture, electronics manufacturing, drug delivery, therapy techniques, diagnostic techniques, space flight and consumer products.

TEXT BOOKS:

| S. No. | Authors | Title | Publishers | Year of publication |
|--------|--------------|----------------------|------------------------------|---------------------|
| 1. | S. Shanmugam | Nanotechnology | MJP Publishers, Chennai | 2010 |
| 2. | S. Balaji | Nanobiotechnology | MJP Publishers, Chennai. | 2010 |
| 3. | T. Pradeep | Nano: The Essentials | Tata Mc-Graw Hill, New Delhi | 2007 |

REFERENCE BOOKS:

| S. No. | Authors | Title | Publishers | Year of publication |
|--------|--|---|-----------------------------|---------------------|
| 1. | Wilson, Kamali Kannangara, Geoff Smith, Michelle Simmons, Burkhard Raguse and Mick | Nanotechnology: Basic Science and Emerging Technologies | Overseas Press | 2005 |
| 2. | G. B. Segreev | Nanochemistry | Elsevier, New York | 2006 |
| 3. | S. Sivanesan and J. Nandagopal | A Textbook of Engineering Chemistry-I | VK Publications | 2015 |
| 4. | P. Charles P. Poole and Frank J. Owens | Introduction to Nanotechnology | A John Wiley and Sons, Inc. | 2003 |

TEACHING METHODOLOGY:

- Conventional chalk and board teaching
- Power Point Presentations
- Assignments
- Animated videos
- Chalk and Board
- Interactive sessions
- To get recent information through the internet.
- Engaging students in cooperative learning.
- Learning through quiz design

SYLLABUS DESIGNER:

- Dr. S. Sashikala, Assistant Professor of Chemistry