Reg.No:						

# D. K. M. COLLEGE FOR WOMEN (AUTONOMOUS), VELLORE-1 SEMESTER EXAMINATIONS

## **JUNE - 2022**

**21CCH2A** 

### **GENERAL CHEMISTRY - I**

Time: 3 Hours Max. Marks: 75

## SECTION – A $(10 \times 2 = 20)$

#### **Answer ALL the questions.**

- 1. Define aromatization.
- 2. How will you prepare acetylene by dehalogenation?
- 3. How will you prepare cyclopentane by pyrolysis of Ca salt of dicarboxylic acid?
- 4. How will you prepare alkynes by oxidation with chromic acid?
- 5. Define electronegativity.
- 6. Calculate the bond order of CO.
- 7. Write a note on the general characteristics of s block elements.
- 8. Write a note on crown ethers.
- 9. Define heat of a reaction.
- 10. What do you know about Boyle temperature?

#### SECTION – B $(5 \times 5 = 25)$

#### **Answer ALL the questions.**

11. (a) Explain any two methods for the preparation of alkanes.

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- (b) What are dienes? How will you classify dienes? Discuss their stability.
- 12. (a) Write notes on acetylides.

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- (b) Discuss the substitution and ring opening reactions of cycloalkanes.
- 13. (a) Define lattice energy. Explain Born Haber cycle.

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- (b) Draw the MO diagram of O<sub>2</sub> molecule.
- 14. (a) Explain the anomalous behaviour of Lithium.

(Or)

- (b) How would you extract beryllium? Write the uses of Be.
- 15. (a) How will you calculate  $\Delta H$  from  $\Delta E$ ?

(Or

(b) Discuss Maxwell's distribution of molecular velocities.

## SECTION – C $(3 \times 10 = 30)$

#### Answer any THREE of the following questions.

- 16. (a) Define chlorination. Explain the mechanism of free radical substitution reaction.
  - (b) Write notes on Markonikkof's rule.
- 17. Write note on the following:
- (i) Wurtz reaction
- (ii) Dieckmann's condensation
- (iii) Preparation of cyclohexane by cycloaddition.
- 18. Based on VSEPR theory, predict the geometry of H<sub>2</sub>O, BF<sub>3</sub> and SF<sub>6</sub> molecules.
- 19. Write a notes on the extraction of Magnesium, its physical and chemical properties and uses.
- 20. (i) Discuss the variation of heat of reaction with temperature.
  - (ii) Derive an expression for Joule-Thomson coefficient.